

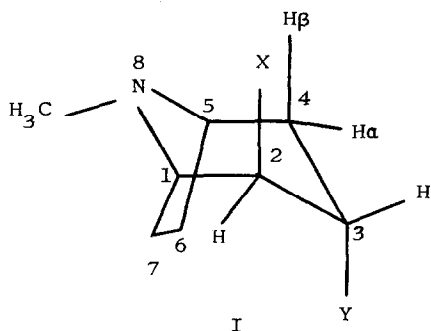
CONFORMATION OF DISUBSTITUTED TROPANES

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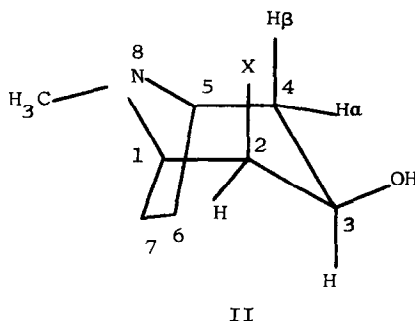
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Recent interest in the conformations of tropane and piperidine derivatives (1-9) prompts us to report our interpretation of the NMR spectra (10) of three tropanes that are substituted in the 2- and 3-positions (11). The chemical shift data listed in Table I has two particularly significant features. The absorption of the hydrogens at the 6- and 7-positions in Ia and



- a: X = Y = Br
b: X = Br; Y = OH
c: X = H; Y = OH



- a: X = H
b: X = Br

Ib as well as in tropine (3- α -hydroxytropine, Ic) (3) is shifted downfield relative to the absorption in IIb and in pseudotropine (3- β -hydroxytropine, IIa). This deshielding of the hydrogens on the ethane bridge indicates (3) proximity of the electronegative 3- α -substituent and is possible only in a chair conformation. Similarly, the β -hydrogen on the 4-position ($H_{4\beta}$) is deshielded relative to the α -hydrogen ($H_{4\alpha}$) in compounds Ia, Ib, and IIb

but not in the compounds lacking the 2-substituents. This deshielding, from the 1,3-diaxial proximity of $H_{4\beta}$ and the 2β -bromine, is evidence of the chair-like conformation of the three disubstituted tropanes.

Table 1

Chemical Shifts (ppm from TMS)							
	H_1H_5	H_6H_7	H_2	N- CH_3	H_3	$H_{4\beta}$	
Tropine (3)	3.0	2.05	1.6-1.9	2.16	3.9	---	---
Pseudotropine (3)	3.07	1.5-1.8	1.5-1.8	2.22	3.7	---	---
Ia	3.25	2.25	4.60	2.31	4.49	2.77	2.05
Ib	3.15	1.99	4.05	2.23	--(a)	2.27	1.55
IIb	3.37	1.65	4.30	2.22	--(b)		

(a) The H_2 and H_3 protons were indistinguishable.

(b) This proton is believed to be hidden under the H_1, H_5 multiplet.

A somewhat better approximation of the conformations of the disubstituted tropanes is available from a consideration of the various coupling constants. The ABX system comprising $H_{4\beta}$, $H_{4\alpha}$, and H_5 was investigated using decoupling techniques by irradiation at H_3 . The other pertinent coupling constants were obtained from the uncoupled spectra. Examination of the J values in Table II shows that all of the coupling constants are of the magnitude expected for axial-equatorial and equatorial-equatorial couplings. However, only the cis-bromohydrin IIb shows the identical coupling constants expected (12) from a perfect chair conformation. The inequality of $J_{4\beta-5}$ and $J_{4\alpha-5}$ as well as $J_{4\beta-3}$ and $J_{4\alpha-3}$ observed for both Ia and Ib indicates that these compounds exist in a distorted chair conformation.

All of these data are consistent with compounds Ia and Ib existing in a chair conformation distorted slightly by interactions between the 3β -substituents and the ethane bridge. When this interaction is removed, as in IIb the distortion is removed (13).

Table II
Coupling Constants (Hz)

	$J_{3-4\beta}$	$J_{3-4\alpha}$	$J_{4\beta-5}$	$J_{4\alpha-5}$	$J_{4\alpha-4\beta}$	J_{2-3}	J_{1-2}
Ia	4.0	< 1	4.0	3.0	16	< 1	< 1
Ib	4.0-4.2	< 1	4.0	3.0	15	< 1	< 1
IIb						3.5	3.5

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10. NMR and decoupling studies were carried out in $CDCl_3$ (60MHz) on Perkin-Elmer-Hitachi R-20 and JNM-C-60 HL spectrometers.

11. The synthesis of Ia, Ib, and IIb will be the subject of a future paper.
All new compounds gave satisfactory analyses.
12. J. R. Dyer, Applications of Absorption Spectroscopy of Organic Compounds,
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13. X-ray studies reported (1) for pseudotropine indicated that this molecule
exists in a distorted chair form in the solid state.